# ST. JOSEPH'S COLLEGE FOR WOMEN (AUTONOMOUS), VISAKHAPATNAMVII SEMESTERB.Sc. HONOURS CHEMISTRYTime: 2Hrs/WeekCode CH 7253(2)Revised Syllabus under CBCS 2020-21Marks: 50Physical Chemistry Practical – I

# I.

Course Objective: To train students in quantitative estimation using conductometry

## **Course Outcomes:**

On successful completion of this practical course, student shall be able to:

- List out, identify and handle various equipment in Chemistry lab.
- Learn and apply the concepts of electro chemistry in experiments.
- Be familiar with electro analytical methods and techniques which study an analyte by measuring the potential (volts) and / or current (amperes) in an electro chemical cell containing the analyte..
- Learn the procedures of preparation of standard solutions.5. Acquire skills in operation and calibration of instruments.

### **II. Syllabus:**

Total Hours: 30h (2h/week)

- 1) Conductometric titration of Strong acid versus Strong base
- 2) Dissociation constant of weak acid (CH3COOH) by conductometric method.
- 3) Conductometric titration of Weak acid vs Strong base.
- 4) Determination of cell constant
- 5) Acid-catalyzed hydrolysis of methyl acetate
- 6) Determination of partial molar volume of solute –H2O system by apparent molar volume method.

### **III. Co-Curricular Activities**

Mandatory: (Lab/field training of students by teacher: (lab:10+field:05):

1. **For Teacher**: Training of students by the teacher in laboratory and field for not lessthan15 hours on the field techniques/skills of handling conductometric titrations.

2. For Students: Studentshallvisitarelated industry/chemistrylaboratory inuniversities/research organizations/private sector facility and observes the synthetic reactions. Write their observations and submit a hand written fieldwork/project work report not exceeding 10 pages in the given format to the teacher.

3. Max marks for Field work/projectworkReport:05.

4. Suggested Format for Fieldwork/project work: Title page, student details, index page, details of place visited, observations, findings, and acknowledgements.

5. Unit tests (IE).

## **IV. Reference books:**

1. Vogel's Text Book of Quantitative Chemical Analysis, J. Mendham, R. C. Denney, J. D. Barnes and M. J. Thomas, 4th& 6 th Ed. (Pearson Education Asia)